

# **Electrochemical Biosensors: Mechanism and Applications**

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### ARTICLE INFO ABSTRACT

Presently, the extensive variety of biomaterials and their broad range of applications have broadened the scope of these materials. Biomaterials are integrated into biosensing platforms to create devices characterized by high sensitivity and specificity, rapid detection capabilities, portability, cost-effectiveness, and userfriendly operation. The limelight of this review focus on potential role and applications of bio based electrochemical sensors. The eMIP/PB/SPCE-based lactate sensors demonstrate significant promise as point-of-care (POC) devices for detecting sweat lactate. This technology could be adapted for reagent-free detection of a wide range of molecules. We discussed a dual-trigger and reusable electrochemical biosensor for microRNA. It achieved sequential signal amplification by integrating CHA and tripedal walker mechanisms. The tripedal DNA walker was activated through APE1 enzymatic cleavage reactions. Strand displacement was engineered for electrode regeneration and optimizing signal strength. It effectively discriminates the target from other interfering RNAs and enables detection in serum. This study introduces an innovative non-invasive electrochemical biosensor designed for detecting glucose concentration in human saliva. The biosensor utilizes a nickel foam substrate decorated with needle-like CoO nanowires, providing an optimal surface area for nanowire growth. These nanowires form a self-assembled flower-like nanostructure with a highly porous configuration, featuring an impressive BET surface area of 154.3 m2 g-1 and demonstrating excellent catalytic capabilities.

**KEYWORDS:** lactate sensors, eMIP/PB/SPCE, CHA and tripedal walker, CoO nanowires

## Introduction

Biosensors are analytical devices that combine bio cognitive components and signal transduction devices. Taking advantage of the precision and resolution inherent in biological sensors, they have an enormous ability to study a wide range of target molecules, including small entities, proteins, nucleic acids and cells [1].Among biological sensors, electrochemical biosensors, using electrochemical agents for signalling have clear advantages such as easy integration, higher sensitivity, fast detection kinetics and cost-effectiveness. Electrochemical biosensors are analytical devices that combine a biological recognition element (such as enzymes, antibodies, or DNA) with an electrochemical transducer to detect and quantify specific analytes in a sample [2-4] They are widely used in various fields including medical diagnostics, environmental monitoring,

food safety, and bioprocess control.

The key components of an electrochemical biosensor typically include:

- **1. Biological Recognition Element**: This could be enzymes, antibodies, nucleic acids, or whole cells that selectively interact with the target analyte, leading to a measurable signal.
- **2. Transducer**: Converts the biochemical reaction into a quantifiable electrical signal. Common types include electrodes (working, reference, and counter electrodes), which can measure changes in current, potential, impedance or conductance.

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**3. Electronics and Signal Processing**: These components amplify and process the electrical signal generated by the transducer to provide quantitative information about the analyte concentration. Currently, a variety of biologically active agents, including enzymes, antibodies, antigens, microorganisms, cells, tissues, and nucleic acids, can directly function as both bio recognition and transduction components. Nano enzymes, which exploit their strong physicochemical properties and greatly improve electro catalytic performance, have become an important driving force in the development of electrochemical sensors [5-6]. In essence, the amalgamation of nanozymes and electrochemical sensing has permeated diverse domains, spanning food science, healthcare, environmental monitoring, disease diagnostics, and immune analysis.

**Reagentless Lactate Sensing with Prussian Blue and Electro polymerized Molecularly Imprinted Polymers in Electrochemical Bio sensing**: The rapid progress in bio sensing, microfluidics, flexible electronics, and wireless communications technologies has spurred the creation of wearable biosensors for sweat analysis, with a focus on lactate monitoring. An increase in lactate production can result in cellular acidosis and impair muscle function [7-9]. Elevated levels may also indicate ischemic pressure and serve as a biomarker for tissue perfusion progression.



🧏 Lactate 🥥 Prussian Blue 🧧 Polymer (Ppy + PAPBA)

Fig 1: (A) Diagram depicting the fabrication process of the eMIP/PB/SPCE-based sensor; Field Emission Scanning Electron Microscopy (FE-SEM) images showing (B) the bare SPCE, (C) PB/SPCE, and (D) eMIP/PB/SPCE configurations; (E) Characteristic cyclic voltammetry (CV) curves measured in 0.1 M KCl for (a) bare SPCE, (b) PB/SPCE after PB deposition, (c) after eMIP polymerization, (d) post-lactate template removal, (e) eMIP/PB/SPCE after overnight drying, and (f) following 1 mM lactate incubation; (F) Cyclic voltammograms detailing PB electrodeposition, (G) electro polymerization of 50 mM 3-APBA, 100 mM pyrrole, and 10 mM lactate on PB/SPCE, and (H) an illustration depicting the lactate detection mechanism, showing the eMIP partially obstructing electron transfer, with corresponding Differential Pulse Voltammetry (DPV) and Electrochemical Impedance Spectroscopy (EIS) responses

Therefore, detecting lactate in sweat provides a non invasive and convenient method for assessing exercise intensity [10], optimizing athletic training, enhancing performance, and supporting medical diagnosis and monitoring. Electro polymerization provides a rapid and straightforward method for constructing MIP-based electrochemical sensors, enabling direct deposition of the MIP layer onto the electrode surface. However, many MIP-based electrochemical sensors rely on external redox couples (e.g., ferricyanide/ferrocyanide) to detect binding events between electrochemically inactive analytes and the MIP. This dependence may interfere with the interaction between MIP and analyte molecules and complicate on-body applications [11-14]. Therefore,

there is an urgent need to develop reagent-free MIP-based biosensors for straightforward and convenient lactate detection, advancing wearable MIP-based biosensor technology.

Prussian Blue (PB) stands as a prominent electron-transfer mediator in enzymatic biosensors. The sensor incorporates an electrodeposited layer of Prussian Blue (PB) as the internal redox probe, along with an electropolymerized Molecularly Imprinted Polymer (eMIP). The eMIP utilizes 3-APBA and pyrrole as functional monomers for the specific recognition of lactate. The growth of eMIP [15-18], removal of the template, and lactate rebinding processes were monitored and confirmed in real-time using Electrochemical-Surface Plasmon Resonance (EC-SPR). The sensor's efficacy was assessed in spiked 0.1 M KCl solution and artificial sweat (AS) using Differential Pulse Voltammetry (DPV) and Electrochemical Impedance Spectroscopy (EIS).

Creation of a reusable tripedal DNA walker electrochemical biosensor for highly sensitive detection of microRNA-155: Given that microRNAs (miRNAs) are pivotal biomarkers for disease diagnosis and prognosis, there is a growing emphasis on enhancing detection sensitivity through signal amplification strategies in research on detecting low-abundance miRNAs. MicroRNAs (miRNAs) are short endogenous RNA molecules, typically 20–24 nucleotides long, that play critical regulatory roles within cells [20–24]. Extensive research has demonstrated that miRNAs can either inhibit or promote the expression of genes associated with various diseases [25], establishing them as valuable biomarkers in disease classification, diagnosis, and prognosis [26,27]. For instance, miRNA-21 shows increased expression in breast cancer patients, which significantly decreases post-surgery [28]. Meanwhile, miRNA-155 is notably upregulated in several cancers, including breast cancer [29], leukemia [30], thyroid cancer [31], gastric cancer [32], colorectal cancer [33], bladder cancer [34], and renal cell carcinoma [35]. This intimate connection between miRNAs and cancer progression underscores their importance in cancer screening and therapeutic strategies. Various techniques, such as electrochemistry [36,37], fluorescence [38,39], colorimetry [40,41], and electro chemiluminescence (ECL) [42], have been developed for miRNA detection. However, these methods may lack sufficient sensitivity to detect ultra-low levels of miRNA without signal amplification. To bolster detection capabilities, various signal amplification strategies have been integrated into electrochemical analysis. These include catalytic hairpin reaction (CHA) [44-46], hybridization chain reaction (HCR) [47-49], entropy-driven catalytic reaction (EDC) [50–52], as well as enzyme-free and nuclease-mediated signal amplification methods [53,54]. Notably, DNA walkers have emerged as dynamic nano devices responsive to targets, capable of converting chemical energy into mechanical kinetic energy. This enables incremental movement along a track and amplification of signals throughout the walking process.



Fig 2: Diagram illustrating a regenerative electrochemical biosensor featuring a robust cascade amplification method. This includes CHA activation for generating a tripedal DNA walker (TDW) triggered by the target, along with APE1 enzyme digestion-driven amplification. A: Electrode regeneration via S2 strand displacement reaction; B: Bio sensing process for miRNA 155, utilizing S1 strand displacement to maximize current signal changes induced by the target.

This technology for electrode regeneration, based on a strand displacement strategy, allows for the repetitive use of DNA-modified gold electrodes. This presents an innovative method toward achieving an efficient and economical detection system [55]. The novel electrode regeneration technique exhibits excellent scalability, making it suitable for detecting various disease-related RNA targets. This development shows potential for

implementing signal amplification through DNA nanostructures and introduces further analytical strategies for clinical diagnostics.

Electrochemical biosensor for glucose detection in human saliva: Diabetes, a metabolic disorder, poses significant health risks [61,62]. Hence, monitoring glucose levels is crucial for diabetes management. However, many current methods for detecting glucose, such as blood collection from fingertips or veins, are invasive, increasing the risk of infection and causing psychological distress among patients [63,64]. Therefore, there is a strong need to develop safe and painless non-invasive methods for diagnosing glucose levels in human biofluids. Several electrochemical biosensors utilizing glucose oxidase (GOx) have been created for saliva glucose detection [67]. However, research indicates that GOx enzymes possess inherent drawbacks such as instability and susceptibility to environmental factors like pH, moisture, and temperature [68]. Consequently, researchers have explored enzyme-free glucose sensors employing noble-metal and transitionmetal catalysts to overcome these limitations. In contrast, Ni foam (NF) is recognized as an ideal substrate for manufacturing glucose biosensors due to its three-dimensional interconnected microstructure. It is costeffective owing to its abundant availability, large surface area, and excellent conductivity [66–68]. Therefore, growing transition metal oxides directly on NF presents a promising approach to developing glucose sensors that are both cost-effective and high-performing. Cobalt oxide (CoO), a readily available transition metal oxide material, has garnered significant interest in non-enzymatic glucose sensors due to its easy preparation and theoretically significant electrocatalytic activity [69,70].



Fig 3: SEM images of (a) NF and (b, c, and d) CoO NWs/NF with different magnifications.

Despite these advantages, bulk CoO suffers from limited active sites and poor conductivity, which hinders achieving optimal catalytic performance. As a result, efforts have focused on nano structuring bulk CoO to enhance electrocatalytic activity by increasing active surface areas and accelerating reaction kinetics [81,82].

Among nanostructured CoO materials, cobalt oxide nanowires (CoO NWs) offer short radial pathways and efficient electron transport in the axial direction, effectively reducing resistance and enhancing reaction kinetics. In this investigation, we successfully achieved the uniform and extensive growth of mesoporous nanoneedle-assembled CoO microflowers on a nickel foam (NF) substrate. Utilizing this distinctive nanostructure, we developed a non-invasive electrochemical biosensor for detecting glucose levels in human saliva. We systematically evaluated the sensing capabilities of the biosensor.

#### **Conclusion:**

The diverse array of biomaterials and their wide-ranging applications have expanded the possibilities for these materials. Biomaterials are now integrated into biosensing platforms to develop devices known for their high sensitivity, specificity, rapid detection, portability, cost-effectiveness, and user-friendly operation. The tripedal DNA walker is activated via APE1 enzymatic cleavage reactions. Strand displacement is engineered for electrode regeneration, optimizing signal strength and effectively distinguishing the target from other interfering RNAs, allowing for detection in serum. Furthermore, we introduce an innovative non-invasive electrochemical biosensor designed for detecting glucose concentration in human saliva. This biosensor utilizes a nickel foam substrate decorated with needle-like CoO nanowires, providing an optimal surface area for nanowire growth. This technology can be adapted for reagent-free detection across various molecules.

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